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Cells 2: Animal cells



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Chemical reactions 3: Oxidation and displacement

1	Name the product for each of the following reactions	[2 marks]		
	a. copper reacts with oxygen			
	b. iron reacts with oxygen			
2	Complete the following sentences. Use words from the box.	[5 marks]		
	displaced element less metal more oxide oxygen reacted			
	During oxidation, elements react with When a metal is			
	oxidised, it becomes a metal			
During displacement reactions, a reactive element takes the				
	place of a reactive element. The less reactive element			
	is			
3	Identify which of the following are oxidation reactions and which are displacement reactions	[3 marks]		
	a. aluminium + oxygen → aluminium oxide			
Type of reaction				
	b eluminium evide L neteccium - > neteccium evide L eluminium			
	 D. auminium oxide + potassium → potassium oxide + aiuminium 			
	Type of reaction			
	c. potassium + oxygen → potassium oxide			
	Type of reaction			
	Total marks	/10		

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Forces 1: Speed

[1 mark]

2 Write down the equation which links average speed, distance and time.

3 Figure 1 shows three distance-time graphs.



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